UE Coordination and supramolecular chemistry

Niveau d'étude
Bac +4

ECTS
6 crédits

Composante
UFR PhITEM (physique, ingénierie, terre, environnement, mécanique)

- **Langue(s) d'enseignement:** Anglais
- **Ouvert aux étudiants en échange:** Oui
- **Code d'export Apogée:** PAX7NAAB

Présentation

Description

Goal: These lectures will introduce you into the world of coodination chemistry both on a synthetic and a physico-chemical points of view.

Content:

I. General concepts in coordination chemistry

- Metal ions and ligands
- Nomenclature of complexes
- Geometry of complexes with different coordinence
- Isomerism in coordination compounds

II. Thermodynamic and kinetic approaches of complexes in solution

- Formation constants: definition and experimental determination
- Chelate effect, a central concept in coordination and supramolecular chemistry
- Applications to supramolecular recognition of cations
- Inertia and lability, essential kinetic notions for understanding complexes reactivity
- Synthesis of complex dedicated ligands: crown-ethers, Schiff bases, polypyridine, ...

III. Electronic structure of metal complexes

- Counting electrons in complexes: the Green's method
- 16/18 electrons rule
- Reactions implying metal complexes
- Application to homogeneous catalysis
- From crystal field to ligand field
- Construction of Molecular Orbitals diagrams of octahedral metal complexes
- Insight into spectroscopic series

IV. Optical properties of metal complexes

- Spectroscopic terms of metal complexes including lanthanide complexes
- Electronic spectroscopy of metal complexes
- Emission of light by metal complexes

V. Magnetic properties of monometallic complexes

- Origins of the magnetic properties of metal complexes
- Magnetic susceptibility
- From Van Vleck equation to Curie law
- Departures from Curie law
- Spin Cross-Over phenomenon: from definition to applications

Article Analysis

Every student will study and present an article dealing with an application strongly related to the contents of the lecture.

Practical teachings:

Four topics of the lectures will be illustrated during four hours experimental work sessions:

- Synthesis and study of the luminescent properties of lanthanide complexes
- Biomimetic model of molybdic oxo-tranferase enzyme
- Synthesis and properties of a iron(II) spin Cross-Over compound [1]
- Syntheis and study of a mixed-valence compound

To anticipate the Lab work, the practical work is written by each student in a dedicated Labwork notebook [2].

[1] A. Vallée *et al., J. Chem. Educ.* **2013**, *90*, doi: 10.1021/ed4000487

[2] A. Eisenberg *J. Chem. Educ.* **1982**, *59*, 1045.

Heures d'enseignement

UE Coordination and supramolecular chemistry - TP	TP	16h
UE Coordination and supramolecular chemistry - CM	CM	36h
UE Coordination and supramolecular chemistry - Other	Autre	3h

Pré-requis recommandés

Prerequisites: Basics knowledge of general chemistry (electronic structure of the elements, thermochemistry...), in group theory and in coordination chemistry (general definitions, crystal field theory)

Période : Semestre 7

Bibliographie

- Inorganic Chemistry by J. E. Huheey
- Inorganic Chemistry by D. Shriver and P. Atkins
- Supramolecular Chemistry by J. M. Lehn
- Molecular Magnetism by O. Kahn
- Electrons in Molecules by J. P. Launay and M. Verdaguer

Infos pratiques

Lieu(x) ville

Grenoble

Campus

Grenoble - Domaine universitaire